Optical properties of some glassy compositions of the GeSe_{2-x}Sn_x thin films of the ternary chalcogenide system

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ABSTRACT

The optical transmission $T(\lambda)$ spectra has been recorded, in the spectral range of 400-1100 nm, for thermally evaporated Sn doped $GeSe_{2-x}Sn_x$ films. The X- ray diffraction technique was used to confirm the polymeric nature of the studied films. The current optical theories and models were applied to analyze the recorded spectra and to calculate various interesting optical parameters. These include: dispersion of the two components of both complex refractive index (n=n-ik) and complex dielectric constant (ε_1 and ε_2), energy gap (E_g), Urbach energy (E_u), single oscillator energy (E_o), lattice oscillating strength (E_d). At this point, a similar non-monotonic trend is observed for the compositional dependence of various parameters. This has been ascribed to the disorder and/or structural defects introduced due to the incorporation of Sn, up to 33.34 % Sn, in the $GeSe_{2x}Sn_x$ structure.

It is observed that band gap decreases with the increase of Sn concentration in the system. This variation in the band gap is explained on the basis of change in structure of the system due to the introduction of Sn in Ge-Se-Sn glassy system.

Keywords: Chalcogenide glass, Thin films, Optical band gap, Transmission, Urbach energy, Complex dielectric constant, Absorption coefficient.

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الخواص الضوئية لأفلام رقيقة لبعض التركيبات الزجاجية للنظام الشالكوجينى ثلاثى الطور GeSe_{2-x}Sn_x تاريخ الإيداع 2014/05/14 قبل للنشر في 2014/09/23 الملخص

 $Gese_{2,x}Sn_x$ سبّك طيف النفوذية (٨) في المجال الطيفي nm الما00 الأفلام رقيقة من السنمط Gese_{2,x}Sn المحضرة بطريقة التبخير الحراري. ومن ثمّ جرى التأكد من الطبيعة البوليميرية للأفلام المدروسة، المحضرة بطريقة التبخير الحراري. ومن ثمّ جرى التأكد من الطبيعة البوليميرية للأفلام المدروسة، باستخدام تقنية انعراج الأشعة السينية، وكذلك تحليل بيانات طيف النفوذية المسبحل بالاعتماد على المحاذاج والنظريات الحديثة، وذلك بهدف حساب العديد من المتحولات الصوئية التي تتضمن تشتت الماذج والنظريات الحديثة، وذلك بعدف حساب العديد من المتحولات الصوئية التي تتضمن تشتت مركبتي قرينة الاركسار العقدية (n=n-ik)، وثابت العزل الكهربائي (ϵ_1 and ϵ_2)، وعرض الفجوة الطاقية (ϵ_1)، وعاد الكهربائي (ϵ_1)، وعاد الخودي الطاقية (ϵ_2)، وطاقة المديذار العقدية (ϵ_1)، وطاقة المذبذ الأحادي (ϵ_2)، وطقة المديني (ϵ_2)، وطاقة المذبذ الأحادي (ϵ_3)، وطقة المديني (ϵ_2)، وطاقة المذبذ الأحادي (ϵ_3)، وطاقة المذبذ المعد المتحولات العين الشبكي (ϵ_3)، وطاقة المذبذ المديني المعادي الكهربائي (ϵ_3)، وطاقة المذبذ المدين المتحولات الكهربائي (ϵ_2)، وطاقة المذبذ الأحادي (ϵ_3)، وطوق التونية الارتباط غير مطرد لقيم هذه المتحولات بتركيب تلك العينات، وفسر هذا بسبب الطاقية (ϵ_3)، ولا العين المدين المدين المعنون (ϵ_3)، وطاقة المذبذ الأحادي (ϵ_3)، وطوق التندين الشبكي (ϵ_3)، وطاقة المذبين المائين (ϵ_3)، وطاقة المذبين (ϵ_3)، وطاقة المذبين المتحولات بتركيب تلك العينات، وفسر هذا بسبب العموانية و رأ و العيوب البنيوية الناتجة من دخول عنصر القصدير بنسبة أكثر من 33.34% فلي بنيسة المن من 33.34% فلي بنيسة المن من 33.34% فلي بنيسة النامية المالية (ϵ_3).

لوحظ أن الفجوة الطاقية (E_g) تتناقص بزيادة تركيز عنصر القصدير Sn في المنظومة المدروسة، وهذا التغير في عرض الفجوة الطاقية يفسر أيـضاً نتيجـة للتغيـر فـي البنيـة الزجاجيـة للمنظومـة GeSe_{2-x}Sn_x بسبب دخول الـقصدير فيها.

الكلمات المفتاحية. زجاج شالكوجيني، فلم رقيق، الفجوة الطاقية الضوئية، النفوذية، طاقة ايربخ، ثابت العزل، معامل الامتصاص

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1. Introduction

Chalcogenide glasses (ChG's) with their flexible structure, enormous variation in properties, and almost unlimited ability for doping and alloying are promising candidates for photonic applications due to their attractive optical properties, such as high refractive index, high photosensitivity and large optical nonlinearity. The investigation of the optical properties of ChG's is of considerable interest and affords critical information about the electronic band structure, optical transitions and relaxation mechanisms. The optical and electrical properties of ChG's are generally much less sensitive to nonstoichiometry and the presence of impurities is less sensitive than crystalline semiconductors. Moreover, ChG's are conducive for use in fiber optics and integrated optics since they have many unique optical properties and exhibit a good transparency in the infrared region, [1,2]. Because of their IR transparency, photosensitivity, high optical nonlinearity, and rare-earth doping potential, these glasses have been utilized to fabricate photonic devices such as fibers, [3] planar waveguides, [4,5] gratings, all-optical switches, and fiber amplifiers.

Device applications using ChG's usually require the glass to be processed into either fiber or thin film form. The wide practical application of thin amorphous chalcogenide films, especially Seglasses on the basis of Ge, is closely connected with their good transparency in the visible and IR spectral regions and with the possibility to create optical media with defined values of the refractive index, dispersion, and extinction coefficients. The relatively low energy of the chemical bonds in the Ge-based ChG's offers the opportunity for photostructural transformations and a number of other light-induced effects, all of which are typically accompanied by considerable changes in the chalcogenide's optical constants,[6].

2. Theoretical background

2.1) Optical absorption spectroscopy of amorphous semiconductors. In amorphous semiconductors, the optical absorption edge spectra generally contain three distinct region [7]:

a) <u>**High absorption region**</u> ($\alpha \ge 10^4$ cm⁻¹), which involves the optical transition between valence band and conduction band which determines the optical band gap. The absorption coefficient in this region is given by:

$$\alpha(hv) = B(hv - E_g)^P$$

where E_g is the optical band gap and B is a constant related to band tailing parameter. In the above equation, p=1/2 for a direct allowed transition = 3/2 for a direct forbidden transition, p=2 for an indirect allowed transition and p=3 for an indirect forbidden transition.

b) <u>Spectral region with</u> $(10^2 \ge \alpha \ge 10^4 \text{ cm}^{-1})$ is called Urbach's exponential tail region in which absorption depends exponentially on photon energy [8] and is given by:

$$\alpha(hv) = \alpha_0 exp\left(\frac{hv}{E_u}\right)$$

Where α_0 is a constant and E_u is interpreted as band tailing width of localized states, where the absorption in this region is due to transitions between extended states in one of the bands and localized states in the exponential tail of the other band [9]. The inverse slope, or the width of the exponential edge (E_u), reflects the width of the more extended band tail that is often called Urbach energy. Such exponential, or Urbach, edge is usually ascribed to localized states at the band edges. In other words, the Urbach edge is determined by the degree of disorder (e.g., charged impurities) and/or structural defects (e.g., broken or dangling bonds, vacancies, non-bridging atoms, or chain ends) in the considered semiconductor material [10]. The structural defects are collected together or assembled with impurities thus forming defect complexes in order to obtain a lower energy state. c) **The region with** ($\alpha \ge 10^2$ cm⁻¹) involves low energy absorption and originates from defects and impurities.

3. Experimental procedure

Homogeneous massive bulk ingots (5gr) of $GeSe_{2-x}Sn_x$ (where x = 0, 0.2, 0.4, 0.6, 0.8 &1) glassy materials were prepared by melt

quenching technique[11]. The high purity constituent materials were taken in elemental powders form. Mixture of these powders was then sealed in a quartz ampoule and this ampoule was heated at a maximum temperature about 950 C° for 12 hours. The ampoule was frequently shaken to achieve better homogeneity. Finally the heated ampoule was quenched in ice-cooled water. To avoid fracture of the tube and glass ingot, and to minimize inner tension induced by a quenching step the ampoules were subsequently returned to the furnace for annealing for 5 h at 40 C° below the glass transition temperature T_o. So obtained ingot of material was grinded into its powder form. These powders were then used to deposit films using thermal evaporation technique under a starting pressure of 10^{-5} Torr on glass substrates (the substrate temperature $\approx 250 \text{ C}^0$, rate of deposition \approx 7-10 A^o.s⁻¹, The thickness of thin films \approx 400 nm). The films were kept inside the deposition chamber for 10 hours to achieve the metastable equilibrium. The nature of the structural phase of the as-prepared films were confirmed using an X-ray diffraction (XRD) computerized system (model: Ital products ADP 200 diffractometer) with Cu, Ka radiation of wavelength $\lambda = 0.15406$ nm, Fig(1). Measurements of the transmittance $T(\lambda)$ were carried out using a computer-aided UV-VIS-NIR spectrophotometer (type: OPTIZEN 2120UV PLUS). The function $T=f(\lambda)$ was recorded at normal incidence. A resolution limit of 0.3 nm and a sampling interval of 1 nm are utilized for the different measuring points. Also, the accuracy of measuring $T(\lambda)$ is 0.004. The measurements were carried out at RT for the recorded entire spectral range of 300–1100 nm.

4. **Results and discussion**

4.1. Nature of as-deposited films

The XRD pattern shown in Fig(1) reveal the amorphous nature of $GeSe_{2-x}Sn_x$ films. Since no peaks were observed in the XRD pattern.



Fig. 1. XRD pattern of $GeSe_{2,x}Sn_x$ glassy film. (where x = 0, 0.2, 0.4, 0.6,0.8 &1).

4.2. Transmission spectra.

The transmission spectra in the wavelength range 300-1100 nm at normal incidence of the as-evaporated $GeSe_{2-x}Sn_x$ thin films are shown respectively in Fig(2). The averages of the transmission values of all the layers in the transparency region (800 - 1100 nm) are 60 % and 97 % respectively.

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Fig. 2. Transmittance versus wavelength as recorded for the studied GeSe_{2-x}Sn_x films.

4.3. Determination of optical constants

4.3.1. Short-wavelength absorption

Fig(3) shows the dependence of short-wavelength absorption edge $GeSe_{2-x}Sn_x$ system. The short-wavelength absorption edge shifts to the longer wavelength with the addition of Sn content which is induced by two factors. First, it is due to the decreasing relatively strong Ge-Se bonds (49.5 kcal/mol) And the increasing weak Ge-Sn bonds (37.4 kcal/mol). Second, the addition of tin will generally decrease the optical band gap (Eg) [25-27], These two factors result in a red-shift of the short-wave absorption edge.



Fig. 3. The calculated absorption coefficient as a function of photon energy in the Tauc and Urbach regions for the investigated films.

4.3.2. Optical band gap

Plot of the function α =f (hu) for each investigated composition that shown in Fig. 4, could be subdivided into two regions :

ü The first region is for the higher values of the absorption coefficient, namely for $\alpha(hv) > 10^4$ cm⁻¹, that corresponds to transitions among extended states in both valence and conduction bands. The functional dependence of the optical absorption coefficient $(\alpha hv)^{1/2}$ versus hv is shown in Fig (4) for the investigated films. For each composition, the energy gap (E_g) has been calculated by fitting the straight part in the high energy region of the function $(\alpha hv)^{1/2} = f(hv)$, locally point by point, to a linear regression line with fitting parameter $R_2 \approx 0.99$ and the results are given in Table 1. At this point, the value of E_g shows a decrease from 2.149±0.013 (This value is in good agreement with previously reported data [24]) to 1.973±0.011 eV with the introduction of 33.34 % Sn into *GeSe*₂.

The decrease in optical band gap may be explained on the basis of the average bond energies of the system. Pauling proposed [13] that single covalent bond energy of heteroatomic bonds D(A-B) can be estimated from the single covalent bond energy of homoatomic bonds D(A-A) and D(B-B) and the electronegativity, χ_A of atom A and χ_B of atom B using the formula:

$$D(A-B) = [D(A-A)D(B-B)]^{1/2} + 30(c_A - c_B)^2$$

The D(A-A) values used are in units of kcal/mol 37.6 for Ge, 44 for Se and 34.2 for Sn, the χ values of the atoms involved are 2.01 for Ge, 2.55 for Se and 1.96 for Sn. The introduction of Sn atoms in the system $GeSe_{2-x}Sn_x$ is accompanied by the replacement of Se atoms by Sn atoms forming Ge-Sn bonds, thus decreasing the concentration of Ge-Se and Se-Se bonds. Since the bond energy of Ge-Sn bonds (37.4 kcal/mol) is lower than that of the Ge-Se (49.5 kcal/mol) bonds, the average bond energy of the system decreases, which results in the reduction in the optical band gap. The relationships between optical gap and content of Sn are illustrated in the upper left insert Fig(4).

 $\ddot{\mathbf{u}}$ The second region for $\alpha(hv)$ is for the lower values of α , that is for α (hv) < 10⁴ cm⁻¹, where the absorption in this region is due to transitions between extended states in one of the bands and localized states in the exponential tail of the other band. The inverse slope, or the width of the exponential edge (E_u) , reflects the width of the more extended band tail that is often called Urbach energy. The calculated values of E_u are given in Table 1 as a function of film composition. These values show an increase at the first addition of Sn content with a small concentration around 6.67% in $GeSe_{1.8}Sn_{0.2}$ sample and then the increasing addition of the Sn content at the expense of Se content decreases the E_u values. It is interesting to note that; a limited addition of Sn content (from 0 to 6.67 At %) increases the E_u values (from 0.183 to 0.199 eV). Such decrease in the E_u indicates an increase in the disorder character of $GeSe_2$ due to the introduction of Sn. This may be due to the formation of charged impurities and/or structural defects in the quasi-gap of $GeSe_{1.8}Sn_{0.2}$ than do in that of $GeSe_2$. This may mean that, the *Sn* atoms [of atomic radii = $1.62 A^{\circ}$] will introduce

to the lattice sites of the parent sample $GeSe_2$, which will produce a non-bridging bond that will disturb the equilibrium of the structure. This might cause a distribution in lattice positions of the other atoms Ge and Se. This in turn goes to more structural disorder in the sample. Also, the fact that $GeSe_{1.8}Sn_{0.2}$ (2.103±0.017 eV) has smaller energy gap than $GeSe_2$ (2.149±0.018 eV), implies that the formation of defects in the narrower gap will have more effect [14,15]. Conversely, the increasing addition of Sn content (from 6.67 to 33.34 At %) leads to decreasing the width of the localized tail states to 0.153 eV. Such a decrease of Eu is attributed to the crystallized character of the thermally evaporated film.). This may mean that, the Sn atom concentration should be enough to that the system should regard Sn atoms as heterogeneous nucleation sites. This leads to make the crystallization procedure a more desirable choice, which justifies the slope decrease in the E_u values. Also, the presence of band tail (E_u) that accompanied the localized states in the gap reflects some degree of disorder in the considered semiconductor film.



Fig. 4. Tauk's plots for determining the optical energy gap of indirect transitions for the investigated films.

single oscillator energy (E_0) , dispersion energy (E_d)				
Film Composition	E_g (eV)	E_u (eV)	E ₀ (eV)	E _d (eV)
GeSe ₂	2.149	0.183	4.577	22.341
$GeSe_{1.8}Sn_{0.2}$	2.103	0.199	3.571	11.565
$GeSe_{1.6}Sn_{0.4}$	2.062	0.169	3.982	15.485
$GeSe_{1.4}Sn_{0.6}$	2.021	0.161	3.765	12.245
$GeSe_{1.2}Sn_{0.8}$	1.994	0.155	3.712	12.823
GeSeSn	1.973	0.153	3.621	13.457

Table 1. The indirect optical energy gap (E_g), Urbach energy (E_u), The single oscillator energy (E_o), dispersion energy (E_d)

4.3.3. Average energy gap and dispersion energy from WDD model

According to the single oscillator model proposed by Wemple and DiDomenico [16] the optical data could be described to a very good approximately by the following formula:

$$n^2 - 1 = \frac{E_o E_d}{E_o^2 - (hv)^2}$$

where, n is the refractive index, E_0 is the average oscillator energy gap, and E_d is the so-called dispersion energy. The latter quantity measures the average strength of the inter-band optical transitions. Plotting $(n^2 - 1)^{-1}$ against $(hv)^2$ allows the determination of the oscillator parameters by fitting a straight line to the points. The value of E_0 and E_d can be directly determined from the slope $(E_0.E_d)^{-1}$ and the intercept on the vertical axis, (E_d/E_0) . The values obtained for the dispersion parameters, E_0 and E_d are tabulated in Table 1. As was found by Tanaka [17], the first approximate value of the optical band gap, E_g , is also derived from the Wemple – DiDomenico dispersion relationship, according to the expression $E_g \approx E_0/2$. The values so obtained from this equation are in almost agreement with the values obtained from the Tauc's extrapolation using the value of the absorption coefficient from the T and R measurements.



Fig. 5. Plots of $(n^2-1)^{-1}$ versus $(hv)^2$ for the investigated films.

4.3.4. Refractive Index

The wavelength dependence on the refractive index has been calculated in the range between 200nm to 1100 nm on the $GeSe_{2-x}Sn_x$ films. Using the equations :

 $n = [4R/(R-1)^2 - k^2]^{1/2} - (R+1)/(R-1)$

where R is the reflectance [19], k is the extinction coefficient. Fig.6 shows the variation of refractive index with wavelength of $GeSe_{2-x}Sn_x$ films for different values of composition. It is found from this Figure that the refractive index decreases with the increasing of wavelength of the incident photon. And we can observed the appearance of a peak occurring in dispersion of the refractive index of the films, this was attributed to the rapid change in the optical absorption coefficient in the vicinity of the absorption edge [18, 19].

4.3.5. Extinction Coefficient

The relation between the extinction coefficient (k) (which is calculated from the relation: $k = \alpha . \lambda / 4\pi$) and wavelength for $GeSe_{2-x}Sn_x$ films is shown in Fig. 7. From this Figure it is found that the extinction coefficient takes the similar behavior of the corresponding absorption coefficient. One can deduce from this figure that the extinction coefficient decreased with increasing the wavelength up to ≈ 760 nm due to the high values of the absorption coefficient at this range of wavelength. The extinction coefficients values are similar to that of the absorption coefficients for the same reasons as mentioned before. Our results are in agreement with the results found by Pan *et al.*, [20] and by Sharma *et al.*, [21].



Fig. 6. Plots of the fitted refractive index versus wavelength for the investigated films.



Fig. 7. The extinction coefficient with wave length for GeSe_{2-x}Sn_x film at different composition.

4.3.6. dielectric constant

The Real ε_1 and Imaginary ε_2 Parts Coefficient Figs. 8a and 8b show the variation of the real (ε_1 , which is calculated from the relation, $\varepsilon_1 = n_2 - k_2$ [21,23]) and imaginary (ε_2 , which is calculated from the relation $\varepsilon_2 = 2nk$) parts of the dielectric constant with the wavelength for *GeSe*_{2-x}*Sn_x* films. The behaviors of ε_1 and ε_2 are the same as those of n and k, respectively with the variation of x values. This is due to that the variations of ε_1 mainly depends on the value of n_2 , because of the smaller values of k_2 comparison with (n_2 , while the imaginary part of the dielectric constant mainly depends on k values which were related to the variations of absorption coefficient.



Fig. 8. (a) The real & (b) imaginary part of dielectric constant as a function of wavelength $GeSe_{2.x}Sn_x$ thin films.

4.4. Infrared optical transmission

Fig.9 shows the optical transmission spectra of $GeSe_{2-x}Sn_x$ glassy samples. It is seen from Fig. 9 that few absorption bands occur in the IR spectra due to highly pure Ge and Se elements in our experiments, which means that the impurities in the $GeSe_{2-x}Sn_x$ film are few. (4.12µm) absorption band are observed in the IR spectra according to Se-H impurity [24]. The transmission of the studied glassy samples extends in the MIR with a cut-off edge beyond 5 µm, it is assigned to the intrinsic multiphonon vibrations of Ge-Se. The transmission rate of the $GeSe_{2-x}Sn_x$ glassy samples decrease with increase the Sncontent, which could be explained by the following two factors. One is owing to more Fresbel's losses with increasing content of Sn, because Fresnel's losses depend on the refractive index, and the increasing content of Sn leads to a shift in the higher refractive index forward the IR region. Another is due to the enhanced crystal-forming ability in the presence of the increasing content of Sn, which leads to the increase of scattering loss of glasses.



Fig. 9. The IR transmission spectra $GeSe_{2-x}Sn_x$ thin films

5. CONCLUSIONS

The optical properties of $GeSe_{2-x}Sn_x$ (where x = 0, 0.2, 0.4, 0.6, 0.8 &1) thin films have been measured and calculated from the transmission spectra. An extensive investigation for the increasing addition of Sn, applying the current optical theories and models on the recorded T(λ) spectra, bringing about the subsequent remarks:

- ∨ A non-monotonic trend, whether towards lower or higher wavelengths, for the Sn-increasing concentration of the two components of the complex dielectric constant ($ε_1$ and $ε_2$), dispersion refractive index (n), optical energy gap of indirect transitions (E_g), Urbach energy (E_u), lattice oscillating strength (E_d) as well as for the single oscillator energy (E_o).
- \checkmark The glassy material has high value of refractive index and decreasing optical band gap with increasing content of Sn in the glassy matrix.
- ✓ There is a good consistency which has been found between the results obtained in the present work and those previously reported by other authors for related alloys.
- ✓ Incorporation of Sn into the $GeSe_{2-x}Sn_x$ host matrix introduces new Ge-Sn chemical bonds, with smaller binding energy than that of the Ge-Se bonds, which explains the increase of n, and decrease of the Tauc gap.
- ✓ Lastly, the change in the refractive index after the addition of Sn content, and increase Urbach energy (E_u) for the $GeSe_{1.8}Sn_{0.2}$ sample makes it, an attractive candidate as optical recording medium.

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