2010/07/15 2011/01/24

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PbS

• .PbS . .

PbS PbS : PbS 25nm

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.70nm : .

Growth And Characterization Of Pbs Nanocrystalline Thin Films Deposited On Glass Substrates By Chemical Bath Deposition

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ABSTRACT

In this paper we present the preparation of PbS nanocrystalline thin films using Chemical Bath Deposition (CBD) technique. We have performed this work in order to study the photoconductivity of PbS semi-conductor thin films. The details of the preparation method are described. Thickness of deposited films has been determined using mechanical and optical methods. From the optical absorption measurements we have determined the band gap values. Using the first approximation parabolic bands model and the obtained values of band gaps, we have determined the size of PbS nanocrystallites. Also, we have investigated the electrical and photoelectrical behaviors of the PbS films. Our study shows that the size of PbS thin films nanocrystallites affects the photoconductive properties of the material. Furthermore, investigations show that there are two different sizes of grains located in two different layers, the first one, with grain's size of about 25nm, concerns the part of PbS deposited directly on the glass substrate and the second layer, with grain's size of about 70nm, concerns the PbS deposited on the first layer.

Key Words: PbS, Nanocrystalline Thin Films, Chemical Bath Deposition, Photoconductivity

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1-3µm

[4-10]

.[7-12]		Quantum Cor	nfinement
Polycrystalline			
(Chemical Bath Depos	ition: CBD)		PbS
[13-15]			
:		PbS	
	pH		
Responsivity	Detectivity	:	
	.[3, 9, 10 16, 17]]	
	:		
			-1
		.()
)			-2
/		.(-
			3
	·		-3
	•		

. .[1, 2]

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17

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$[(K_2Cr_2O_7:H_2SO_4; 1:10), HNO_3]$

[Pb(NO₃)₂: 0.011M], [NaOH: 0.109M], [CS(NH₂)₂: 0.044], H₂O

		:	
$Pb(NO_3)_2 + 2NaOH$	\rightarrow	$Pb(OH)_2 + 2NaNO_3$	(a)
$Pb(OH)_2 + 4NaOH$	\rightarrow	Na ₄ Pb(OH) ₆	(b)
Na ₄ Pb(OH) ₆	\rightarrow	$4 \operatorname{Na}^{+} + \mathbf{HPbO_2}^{-} + 3\operatorname{OH}^{-} + \operatorname{H}_2\operatorname{O}$	(c)
$CS(NH_2)_2 + OH^-$	\rightarrow	$CH_2N_2 + H_2O + SH^-$	(d)
$HPbO_2^- + SH^-$	\rightarrow	$PbS + 2OH^{-}$	(e)
T 1 1 1 T T 1	1 1		

 $(Bi(NO_3)_2)$

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(Hydroxylamine Hydrochloride, NH2OH.HCl

[18,19]

.[20]

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0.018M)

(1)

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		(1)
		(°C)
#1	: Pb(NO ₃) ₂ , CS(NH ₂) ₂ , NaOH, H ₂ O	30
#2	+ (NH ₂ OH.HCl 0.018M, Bi(NO ₃) ₂ 5.27×10 ⁻⁵ M)	30
#3	+ (NH ₂ OH.HCl 0.018M, Bi(NO ₃) ₂ 1.05×10 ⁻⁴ M)	30
#4	#3	24

2011	(27)			
	(4) (3))		
20		()	
. 20	7	(pH=13) 100	pН	
Alpha star				-2
Alpha-step	.20nm		160	
: (1053 line:	s/mm)		:	
	(400-1100nm))	Perkin	Elmer
SYSTEM 2000	Perkin Elme	r 5-20 um)	:	FTIR
Reflectivity	(·			
:[21] Beer Lam	ıbert	$T(\lambda)$ Transmit	ttance	R(λ)
$T(\lambda) = (1 - 1)$	$-R(\lambda)\big)\cdot\exp(\lambda)$	$(-\alpha(\lambda)\cdot d)$		(1)
	$\alpha(\lambda) = \frac{1}{d} \cdot \ln \theta$	$\left[\frac{1-R(\lambda)}{T(\lambda)}\right]$		(2)
PbS		(cm ⁻¹) (cm)	α) d T R	:

 $\alpha(\lambda) \quad T(\lambda) \quad R(\lambda)$ (3) (2) (1)

19

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PbS (I-V) .5mm (Gw-Instek, model

(model: PC500) .GDS-2104)

(Gw-Instek, model GPC-3030D) (GRASEBY IR Systems, model 830) .(InfraRed Industries Inc. model 830)

(SNR)

(Lock-in detection) (STANFORD, model SR830) (5)



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(27)

$$\overline{n} = \frac{\text{Optical thickness}}{\text{Geometric thickness}} = 3.9 \pm 0.3$$
(3)

	(nm)	E _{g1} (eV)	λ _{cut-off1} (μm)	D ₁ (nm)	E _{g2} (eV)	λ _{cut-off2} (μm)	D ₂ (nm)
#1	478	1.39	0.89	24	0.51	2.43	64
#2	310	1.46	0.85	24	0.46	2.70	82
#3	325	1.48	0.84	23	0.60	2.07	50
#4	650	1.21	1.02	27	0.46	2.70	82
$\begin{array}{c} :(nm) \bullet \\ :E_{g2} \ (eV), \ E_{g1} \ (eV) \bullet \\ :\lambda_{cut-off2} \ (\mu m), \ \lambda_{cut-off1} \ (\mu m) \bullet \\ : D_2, \ D_1 \bullet \end{array}$							
							:



-2





R

$$[25] \Delta E_g = \frac{\hbar^2 . \pi^2}{2.R^2} \cdot \left(\frac{1}{m_e^*} + \frac{1}{m_h^*}\right) - \frac{1.786 \times e^2}{\varepsilon.R}$$
(5)

 $\Delta E_g = 2.77 \cdot 10^{-35} \cdot X^2 - 3.47 \cdot 10^{-28} \cdot X \quad (X = \frac{1}{R}, \ n = 3.9, \ \Delta E_g \ in \ Joule, \ R \ in \ m)$ $m_h^* \quad m_e^* :$.PbS ϵ

(5)

[25] 10nm

	2011	(27)		
			(2)	
.(25nr	: n)	:	(1.2-1.5 eV, $\lambda_{cut-off} \cong$	• ε 0.9 μm)
.(70)nm)	(($0.45-0.6 \text{ eV}, \lambda_{\text{cut-off}} \cong 2.$	5 µm)
$(\lambda_{\text{cut-off}} = 3 \ \mu m) \ 0.41 \text{eV}$.41eV	160nm	
	DhC		PbS	-3
	PbS bulk		PbS	
(3)	·		(8) (7)	
	PbS (c		:[29]	
2	$\lambda = 2 \cdot \mathbf{d} \cdot \sin(\mathbf{e})$)) λ		(6) θ:
[23]			TADD	

.PbS



(7)			(3)
	[23]	d		
		XRD	TADI)
$d\begin{pmatrix} a \\ A \end{pmatrix}$ Ref.(23)	$d \begin{pmatrix} \mathring{A} \end{pmatrix} (TADD)$	$d\begin{pmatrix} \mathring{A} \end{pmatrix}$ (XRD)	2θ (°) (XRD)	(hkl)
3.426	3.429	3.461	25.716	(111)
2.969	2.969	2.994	29.818	(200)
		2.648	33.817	(210)
2.095	2.099	2.112	42.781	(220)
1.788	1.790	1.799	50.691	(311)
1.715	1.714	1.721	53.187	(222)
1.483		1.593	57.829	(400)
1.361	1.362	1.367	68.591	(331)
1.325	1.327	1.332	70.364	(420)
1.212	1.212	1.216	78.614	(422)

Scherrer

:[29]

 $D = \frac{k \cdot \lambda}{\beta \cdot \cos(\theta)}$

θ

	β
:	[29]
$\beta = \sqrt{\beta_m^2}$	$-\beta_a^2$
	β_{a}

 $\begin{array}{c} & & & D: \\ k & & & \\ & (& &) \\ & (8) & & \\ & & & \beta_m & : \end{array}$

(7)

(200)

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) .(4) λ

(

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					(4)	
	2θ (°)	(°)	(nm)		(nm)	
	29.818	0.245	4	4		
#1	29.678	0.213	5	7	64	
#2	29.925	0.220	5	3	82	
#2Bis	29.731	0.233	4	8	55	
#3	30.442	0.210	5	9	50	
#4	29.686	0.262	3	9	82	
Pb	S	.(50	nm)	(4)	
·					:	
.P (R	•bS)	.(I-V)		(9)	-4
K ₂)			. (R ₁	
5:	5nm	1.8eV	,		20	
.350nm		.2	21nm	(0.7 20	7 μm)	
(PbS	S PbS (PbS	5) .70nn PbS)	n ((10)	(λ PbS)	$\mu_{\rm cut-off} = 2.5 \ \mu m$) 0.49eV







-5

Sensitivity

(R) Responsivity (S) :(F) $R = \frac{S}{F}$ (9)

Specific Detectivity

$$D^{*} = \frac{\left(S_{N}\right) \cdot \sqrt{\Delta f \cdot A_{d}}}{F} = \frac{R \cdot \sqrt{\Delta f \cdot A_{d}}}{N}$$
(10)
$$\Delta f$$

.

N:

$$\tau = \frac{1}{2\pi \cdot F_{\rm C}} = \frac{1}{2\pi \cdot \Delta \nu}$$
(11)
$$F_{\rm C} = \frac{1}{2\pi \cdot \Delta \nu} + \frac{1}{2\pi \cdot \Delta \nu}$$





(6) PbS

			•	
	#1	#2	#3	#4
Fc (Hz)		~ 2000	~ 2000	~ 2000
Tc (µsec)		~ 80	~ 80	~ 80
$D^{*}(W^{-1} Hz^{1/2} cm)$		$3.3 \times 10^{+7} \pm 1\%$	$2.0 \times 10^{+6} \pm 35\%$	$1.23 \times 10^{+8} \pm 3\%$
Responsivity (V/W)		$4.1 \times 10^{+4} \pm 1\%$	$1.9 \times 10^{+4} \pm 35\%$	$1.85 \times 10^{+6} \pm 3\%$

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(11)

(R) .0.99998

R $R = 1.85 \times 10^{+6} \text{ V} \cdot \text{W}^{-1} (\pm 3\%)$.[26-28] (R \approx 10^{+3} - 10^{+6} \text{ V} \cdot \text{W}^{-1})

 D^* $. D^* = 1.2 \times 10^{+8} W^{-1} \cdot cm \cdot Hz^{1/2} (\pm 3\%)$ $\left(D^* \approx 10^{+8} - 10^{+9} W^{-1} \cdot cm \cdot Hz^{\frac{1}{2}} \right)$



-6



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(13)	.((10)	(9))
			(#4)

.(7)

		(7)	
	(μm) λ _{nk}	(W ⁻¹ Hz ^{1/2} cm)	(V/W)
#2	2.8	$2.7 \times 10^{+9}$	$1.6 \times 10^{+4}$
#4	2.8	$1.1 \times 10^{+10}$	$2.7 \times 10^{+4}$







Range of spectral detectivities for PbS (ATO) at 295 K; 2π FOV, 295-K background. (Santa Barbara Research Center.)

34

(14)

.[27] PbS

2011 (27)

[10, 27]

100

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PbS

(8)

$ \frac{D^{*}(\lambda_{\max})}{D^{*}(BB)} = \frac{2.7 \times 10^{+9}}{3.3 \times 10^{+7}} = \frac{1.1 \times 10^{+10}}{1.23 \times 10^{+8}} \\ \frac{D^{*}(\lambda_{\max})}{D^{*}(BB)} = \frac{82}{89} $		#2	#4
$\frac{D^{*}(BB)}{D^{*}(\lambda_{\max})} \frac{3.3 \times 10^{+7}}{B^{*}(BB)} \frac{1.23 \times 10^{+8}}{82}$	$D^*(\lambda_{\max})$	$2.7 \times 10^{+9}$	$1.1 \times 10^{+10}$
$\frac{D^*(\lambda_{\max})}{D^*(BB)}$ 82 89	$D^*(BB)$	3.3×10 ⁺⁷	1.23×10 ⁺⁸
	$\frac{D^*(\lambda_{\max})}{D^*(BB)}$	82	89

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.[27]





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Bismuth			
Chemical Bath Deposition (CBD)			
Conduction Band			
Detectivity			
Electrical Bandwidth			
Electro-optic			
Energy bands		()
Band Gap			
Linear Adjustment (Linear Fit)			
Photoconductivity	()
Quantum confinement			
Responsivity (Sensitivity)		()
Specific Detectivity			
Specific Resistance			
Valence Band			

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